



Process Synthesis from Reaction Networks

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Process Synthesis from Reaction Networks



Paper: An optimisation-based framework for the conceptual design of reaction-separation processes

**Authors: Qingyuan Kong, Nilay Shah
Department of Chemical Engineering, Imperial College London**

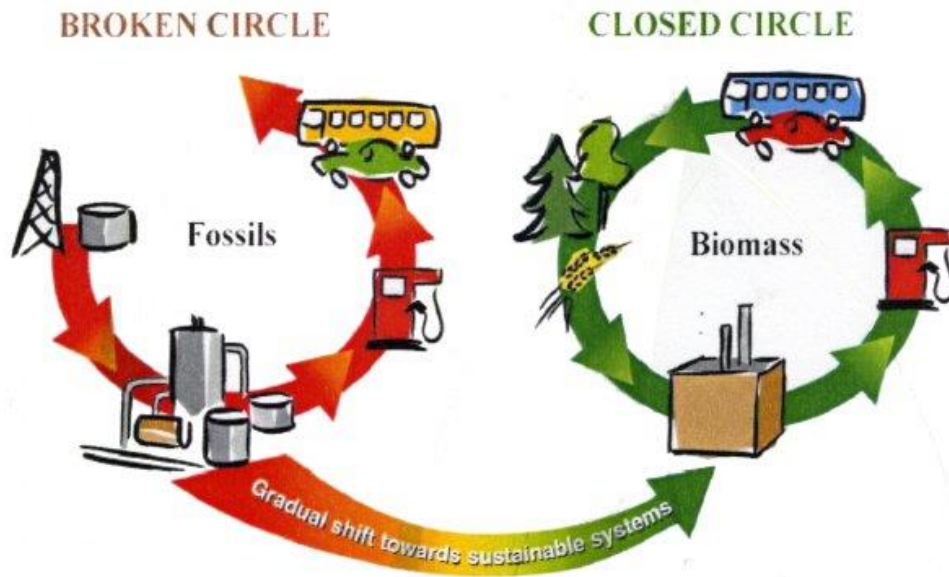


Process Synthesis from Reaction Networks



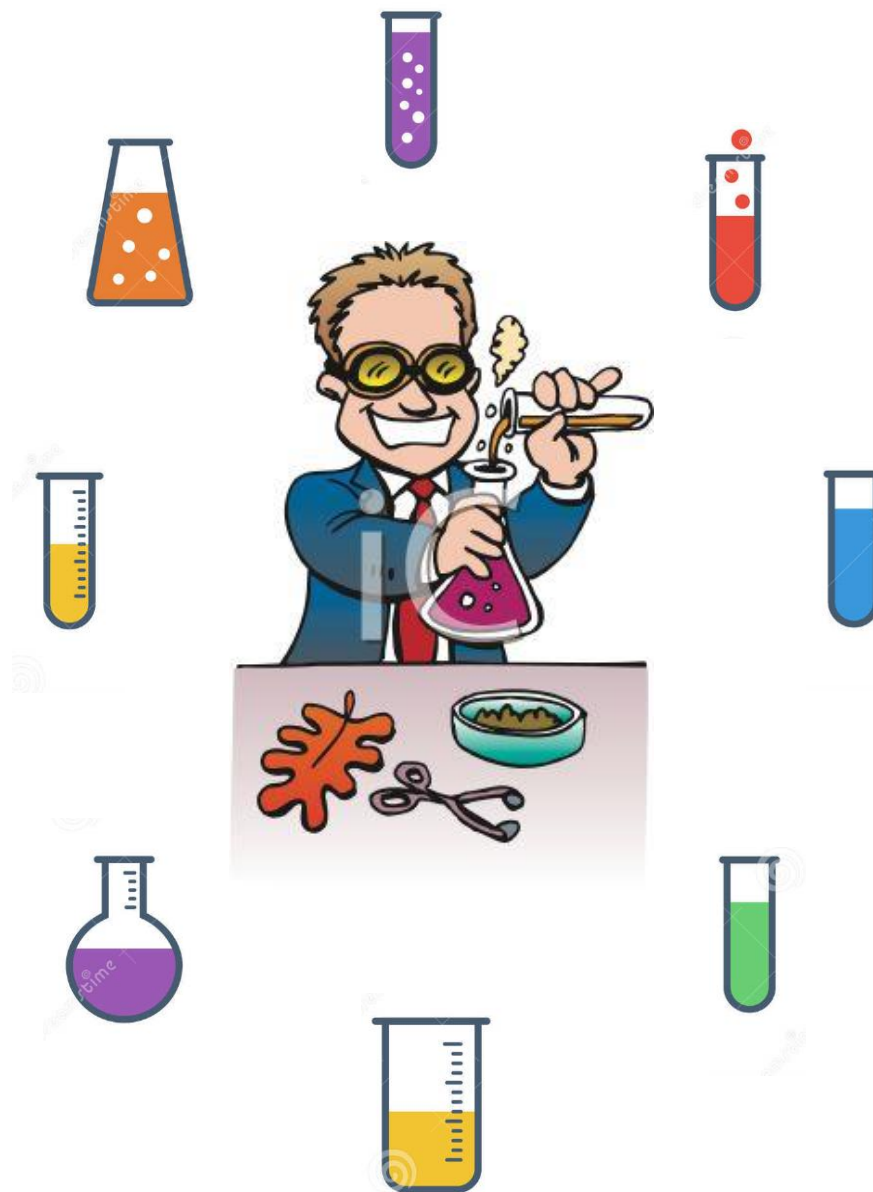
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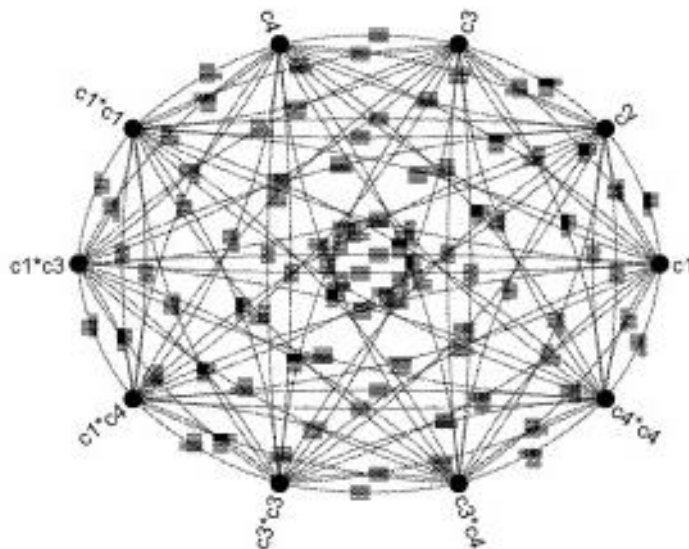


Process Synthesis from Reaction Networks





Reaction Networks



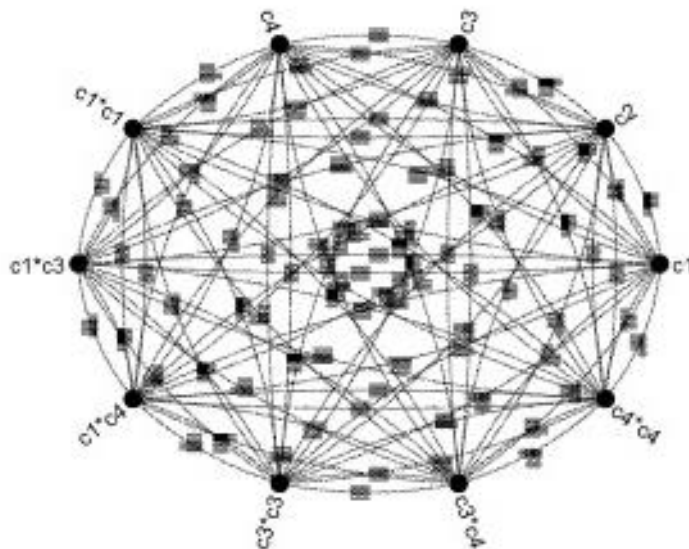
-Literature search

-Speculative reactions

-Automated tools for reaction network generation



Reaction Network Data



-Yields, conversions, etc.

-Pure component property data



Process Synthesis from Reaction Networks?





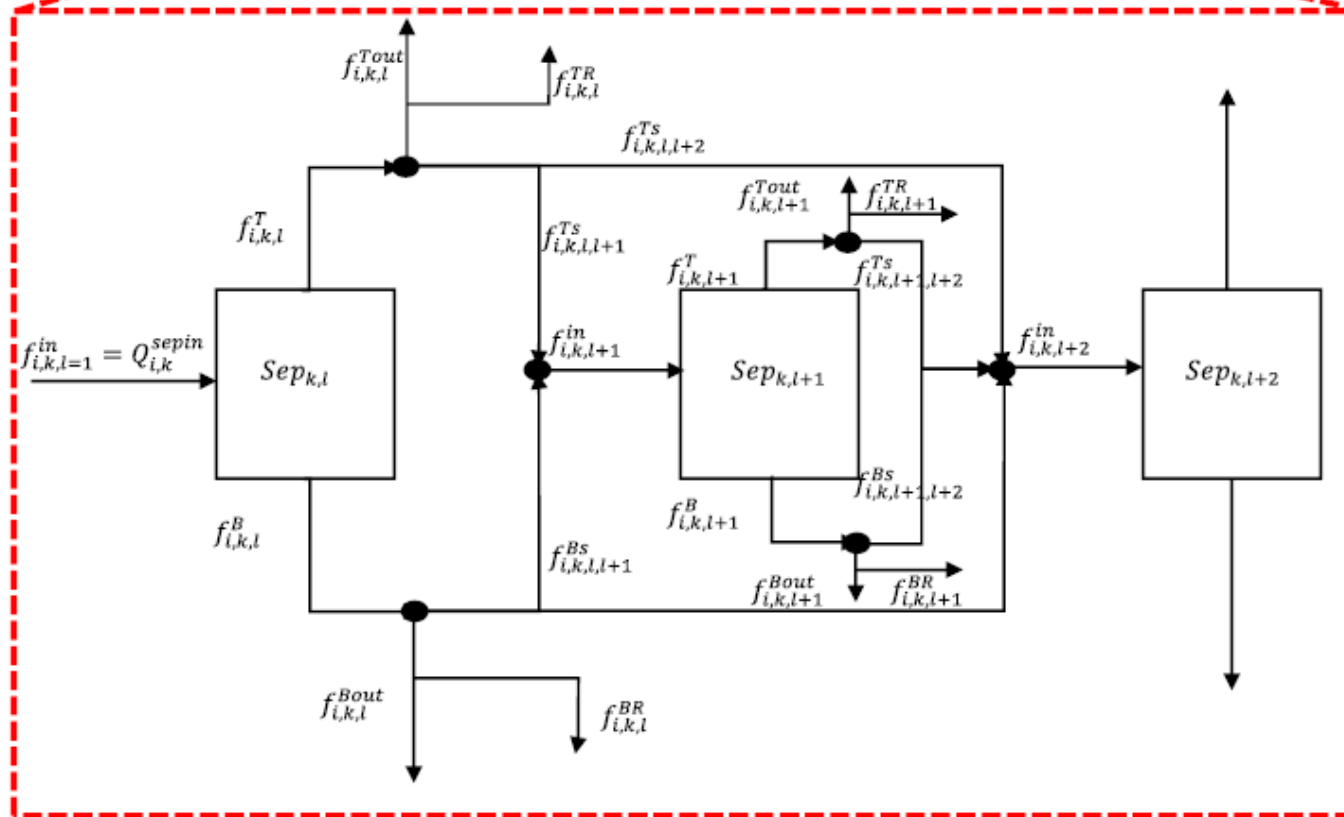
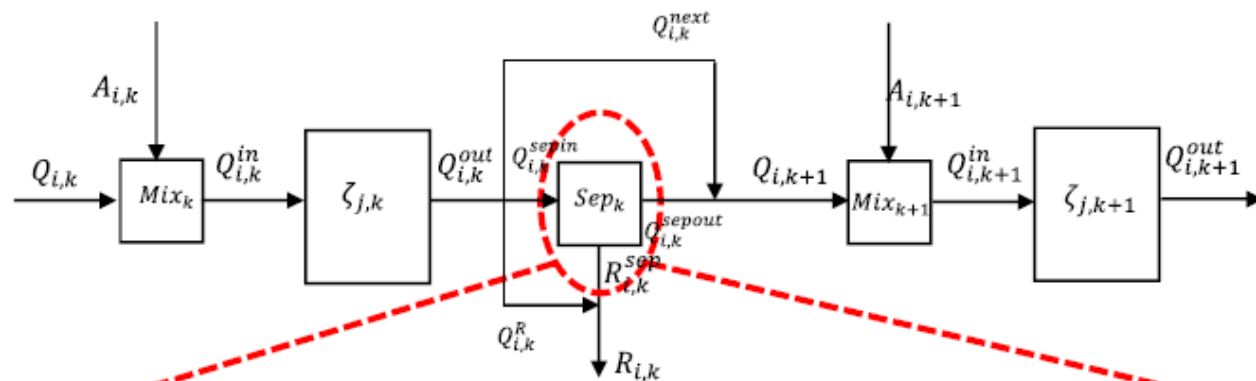
Assumptions



- Conversion and selectivity of each reaction, as well as the boiling point and market value of each component are either known or estimated**
- Distillation (sharp-split) is the method for separation**
- Process operates at steady stage**
- Reactor capital cost is assumed to be negligible**
- A simple overall cost model based on the difficulty of separation is used**



Superstructure-based Methodology

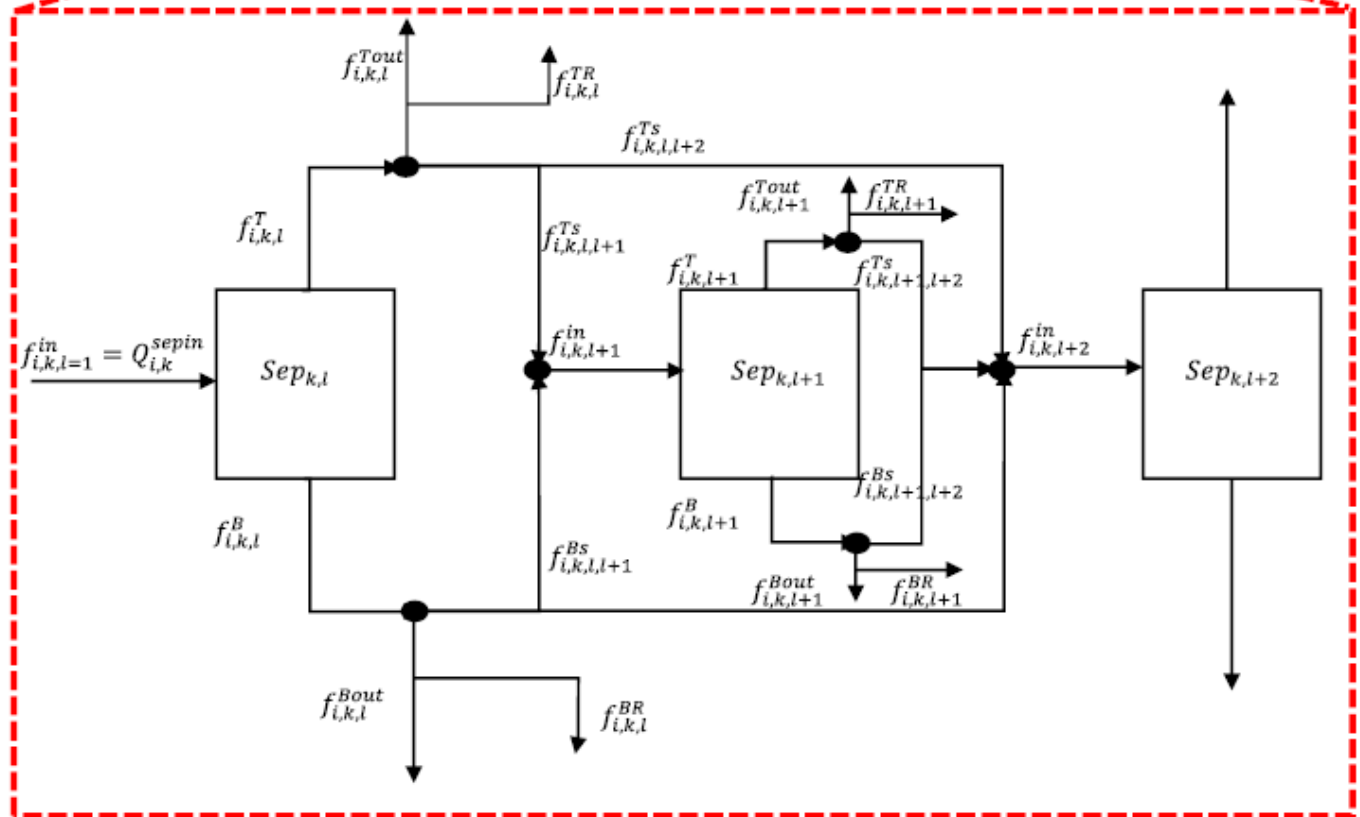
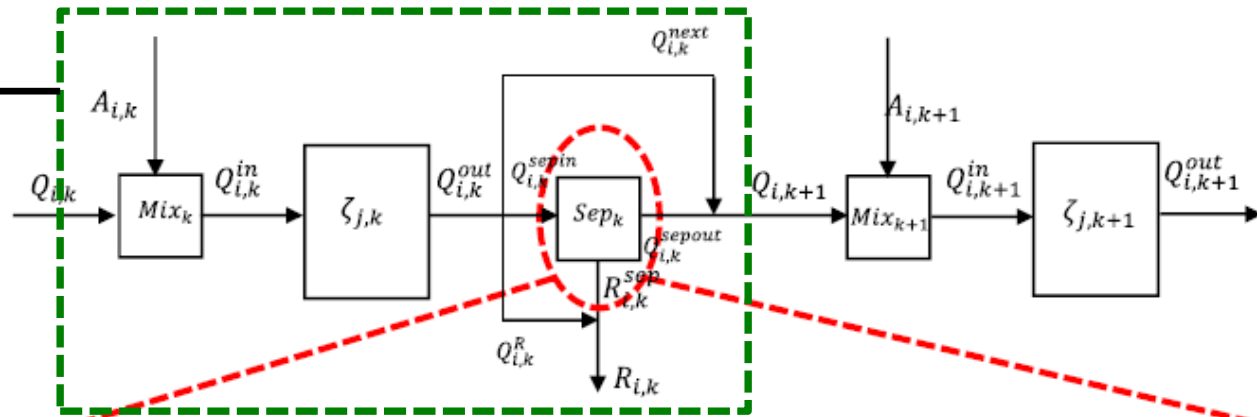




Superstructure-based Methodology

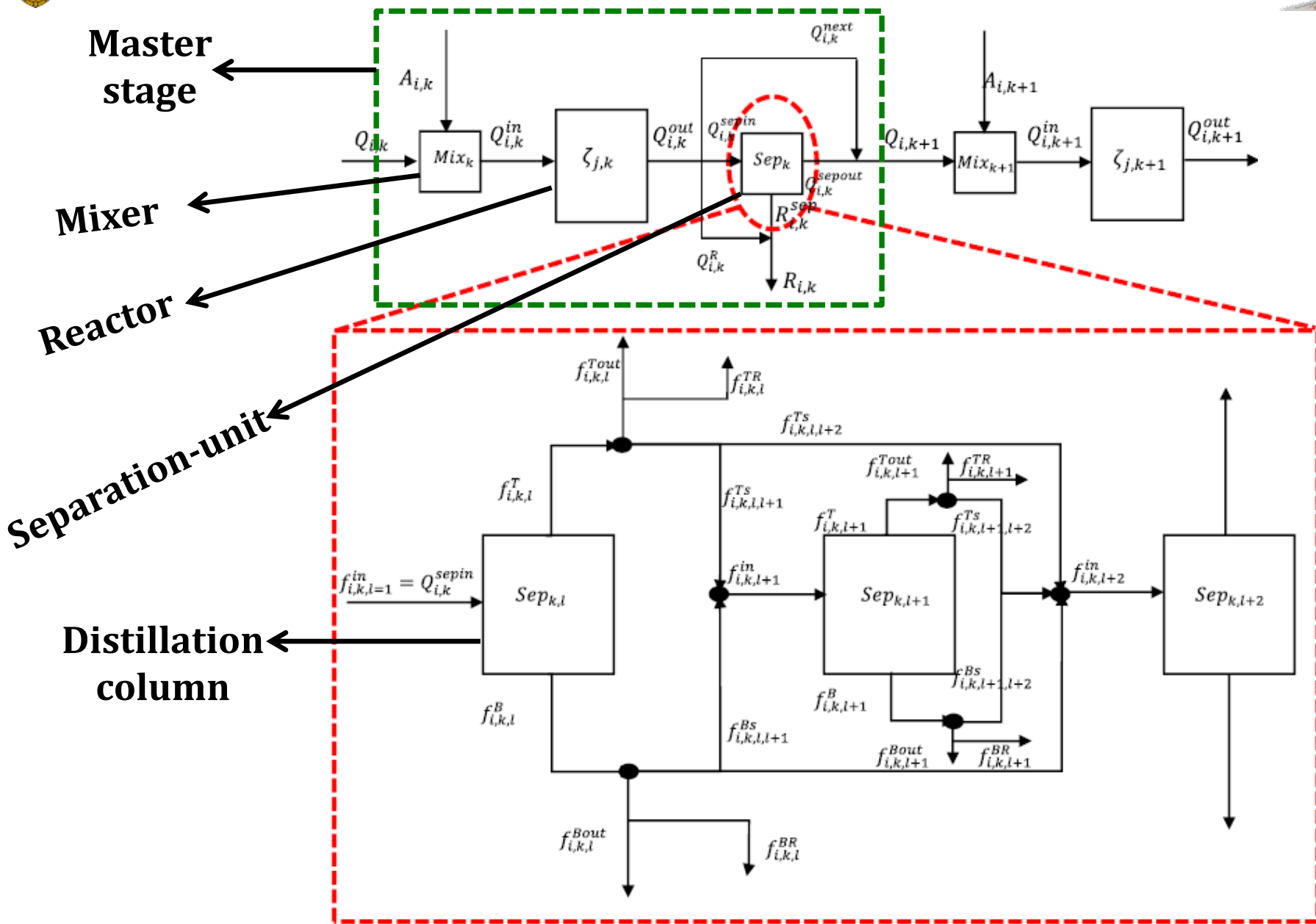


Master stage



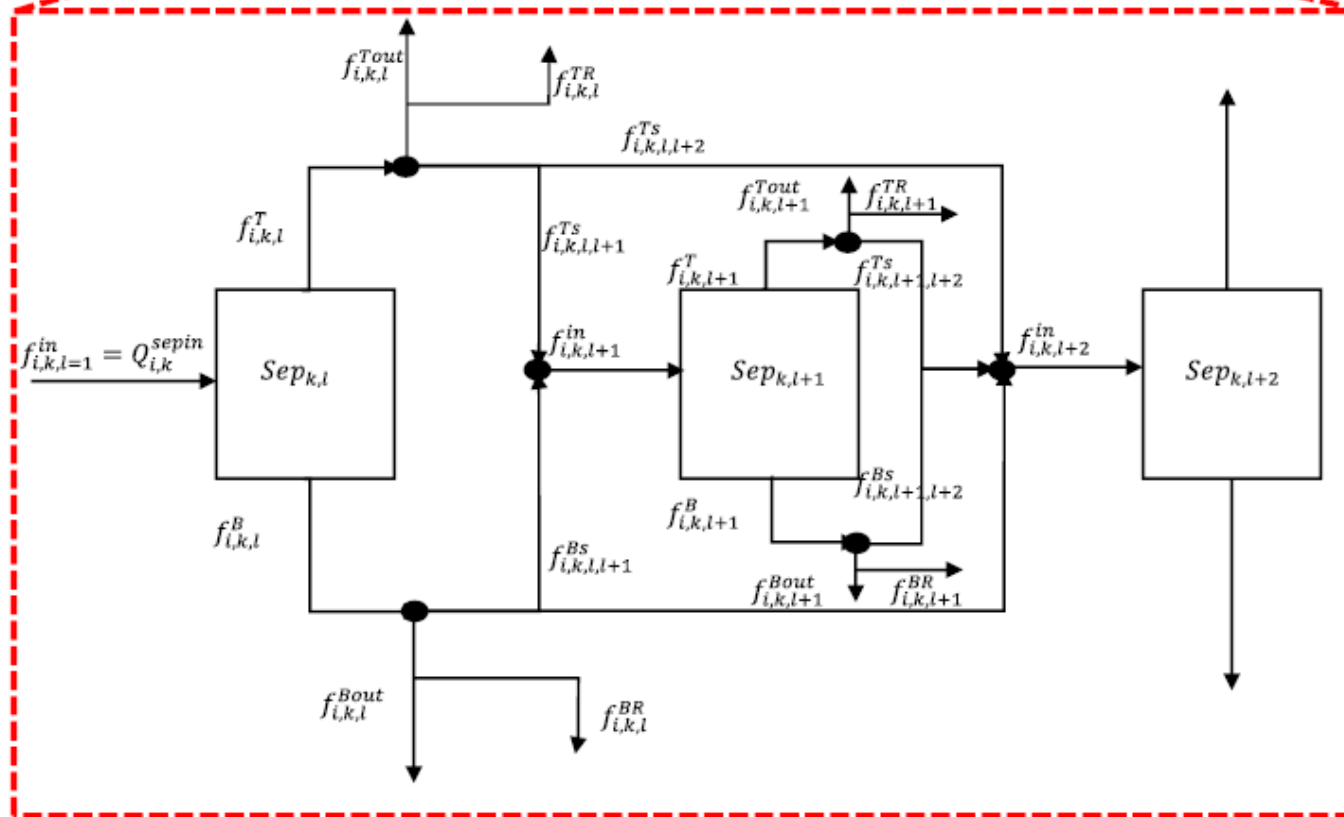
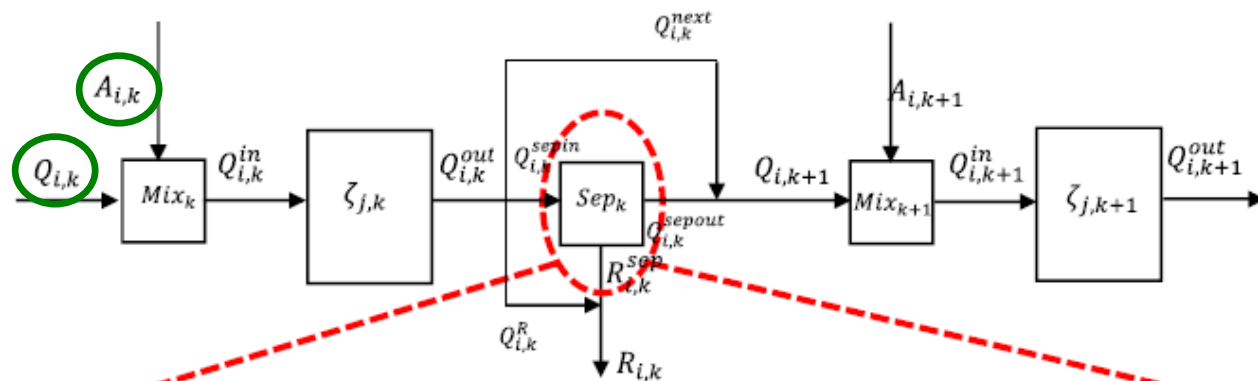


Superstructure-based Methodology



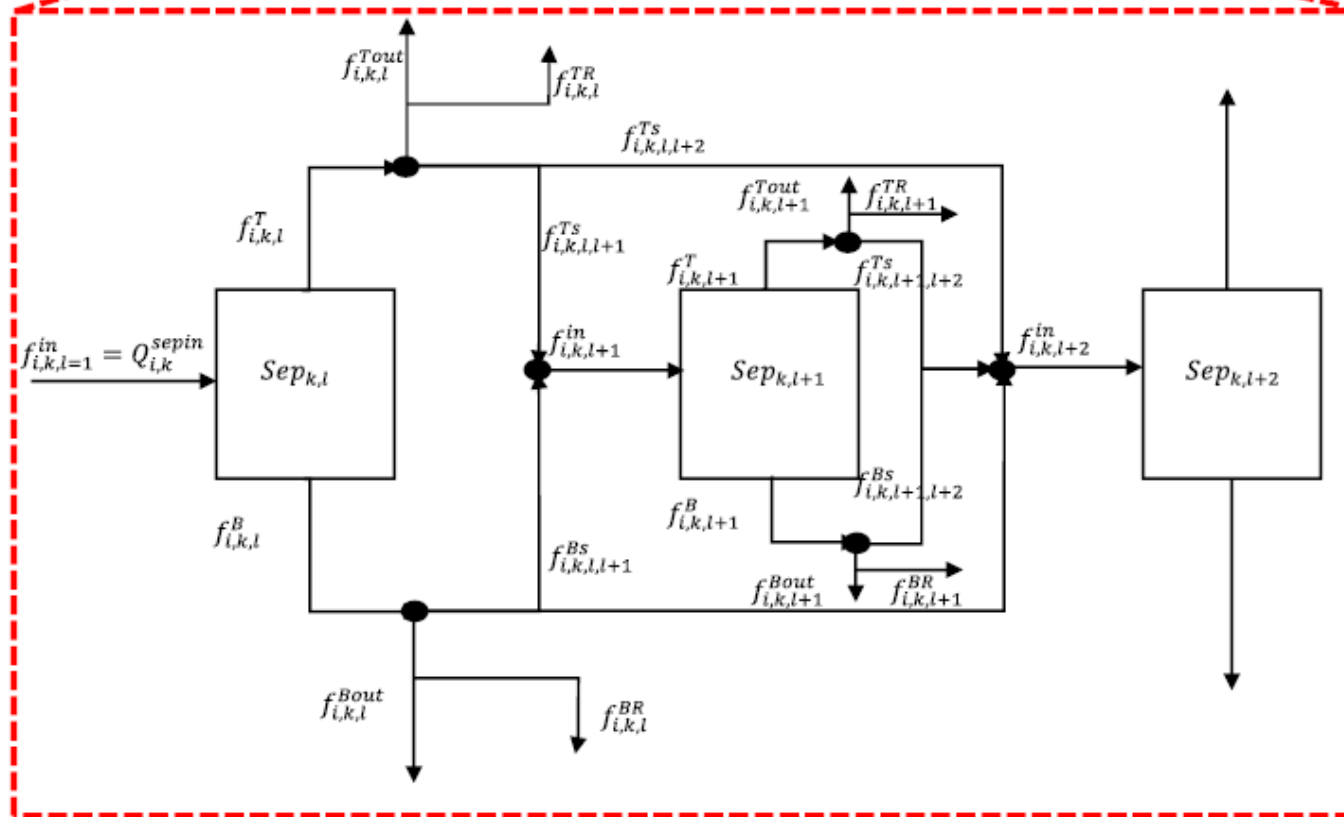
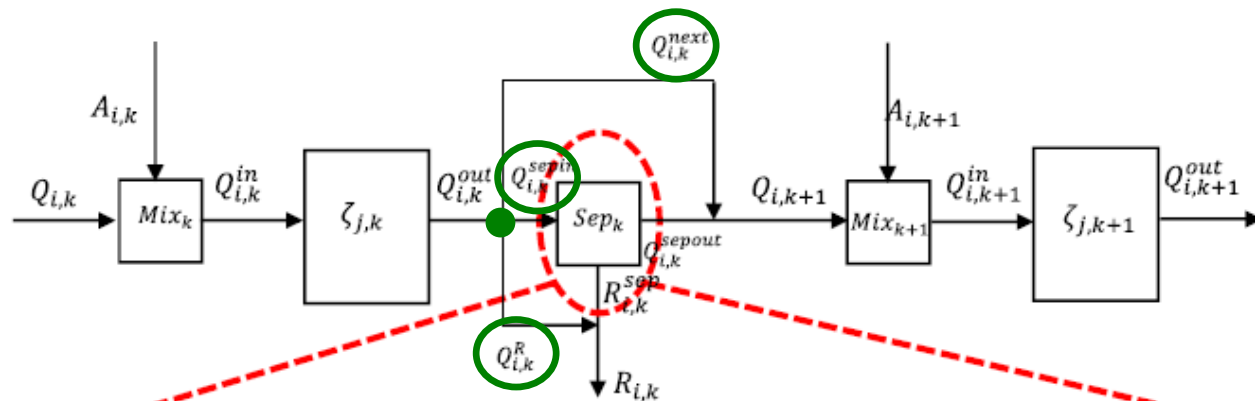


Superstructure-based Methodology



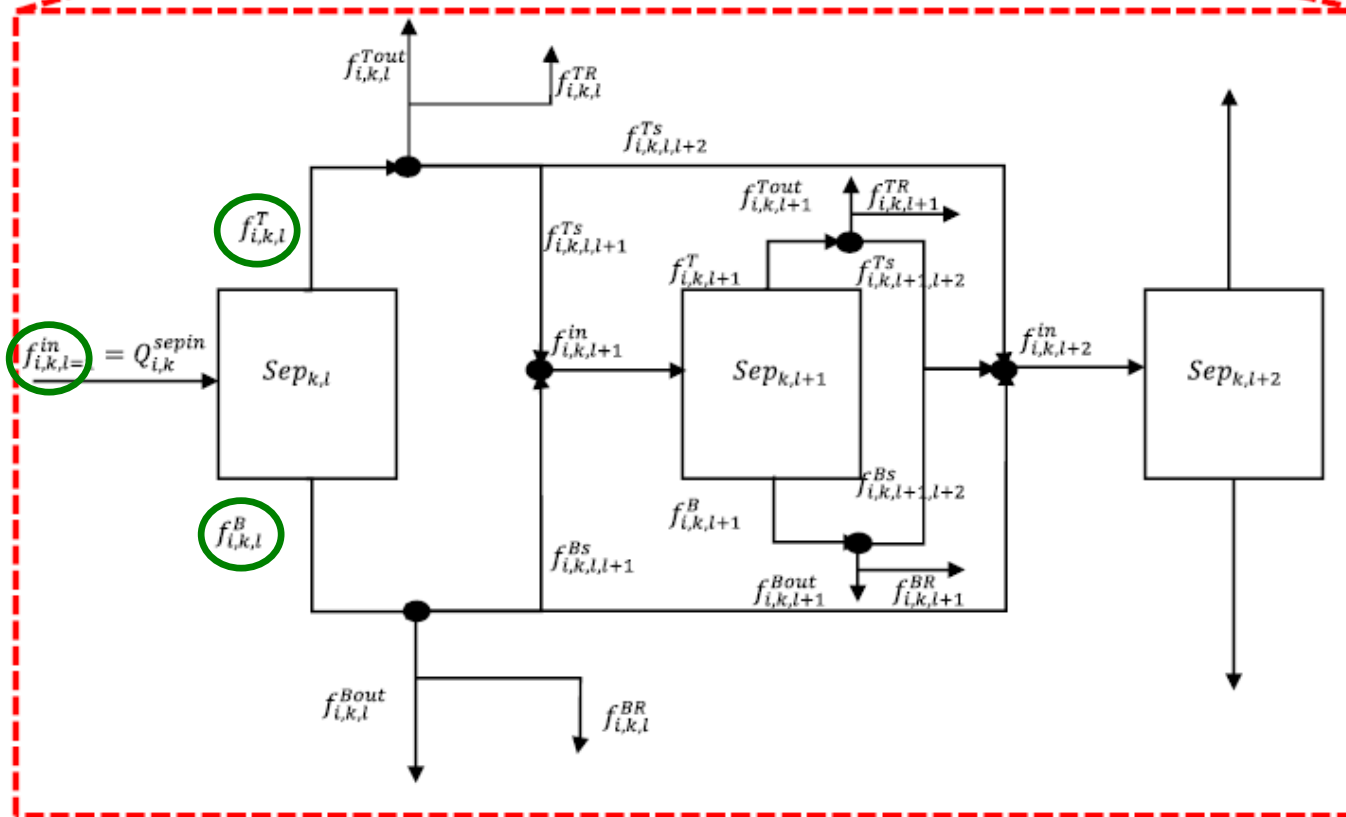
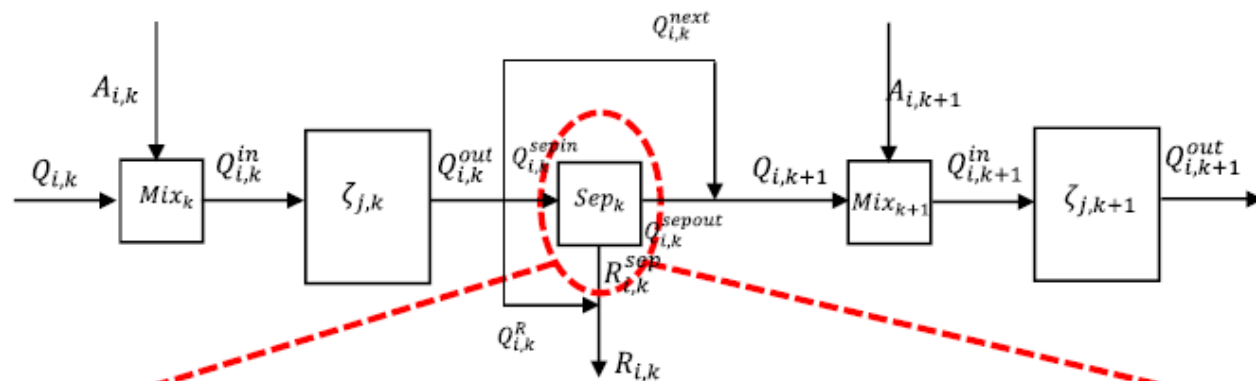


Superstructure-based Methodology



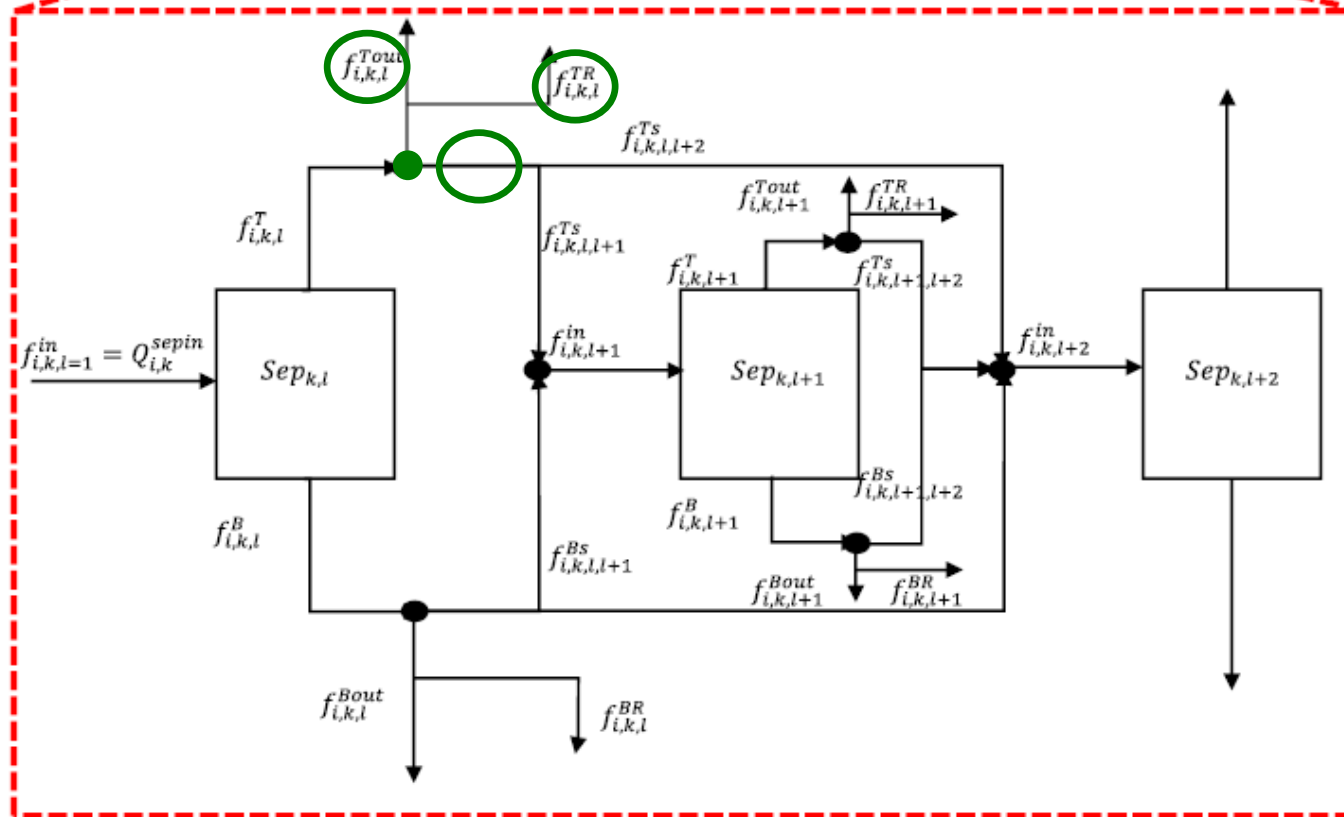
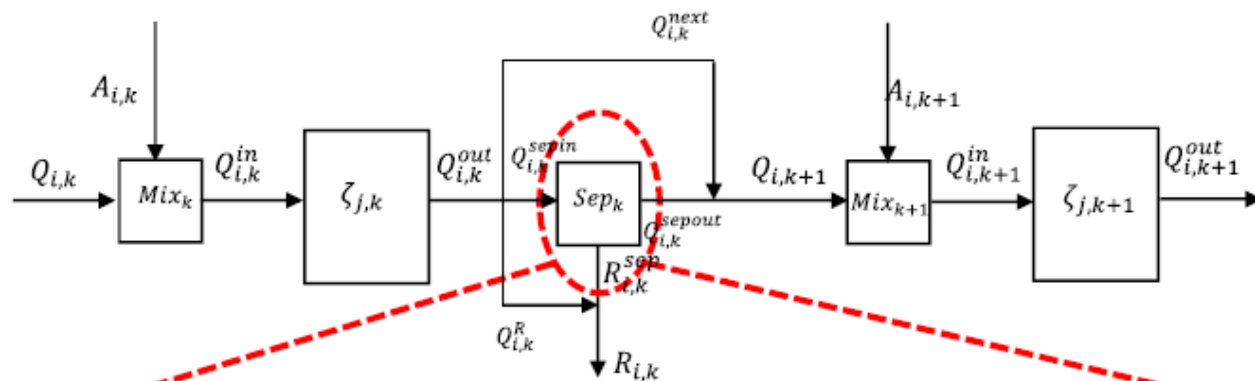


Superstructure-based Methodology



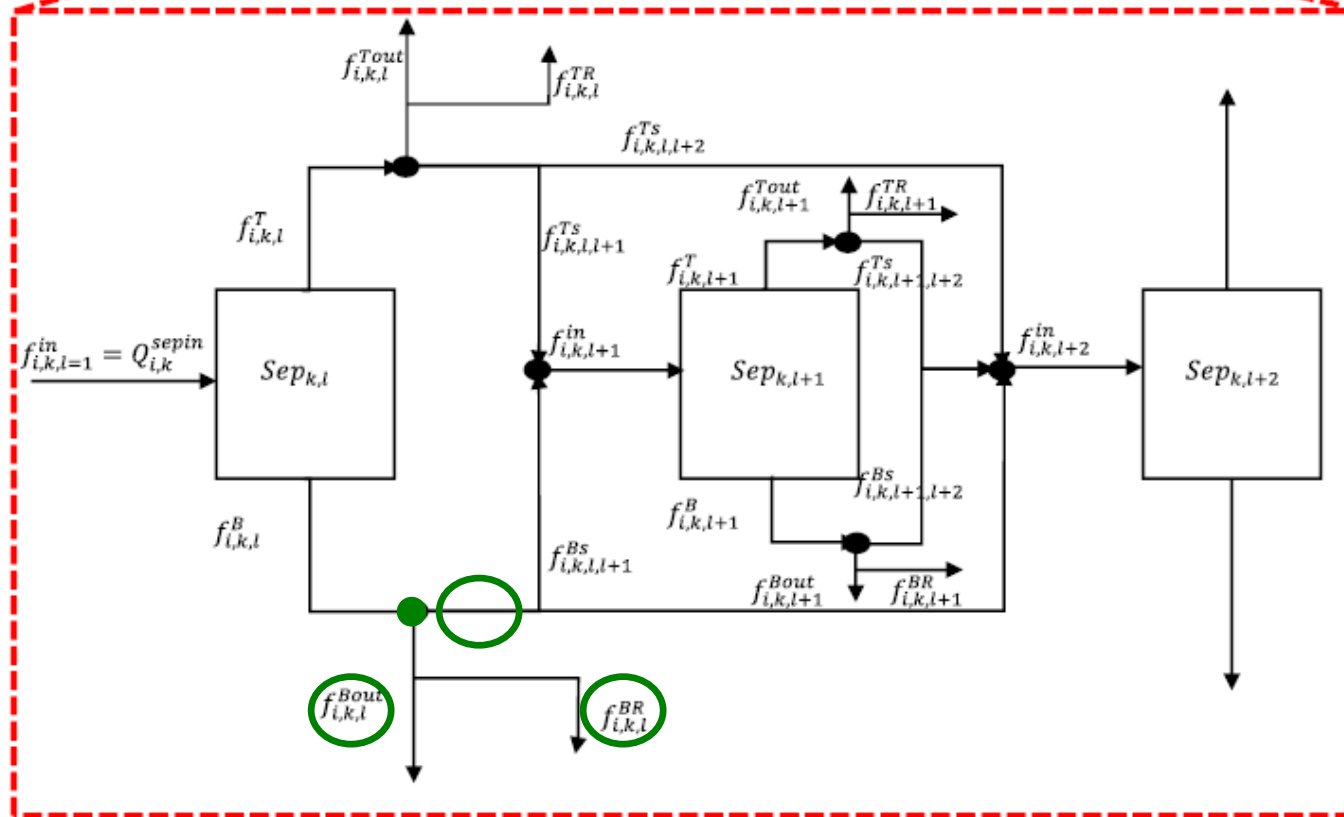
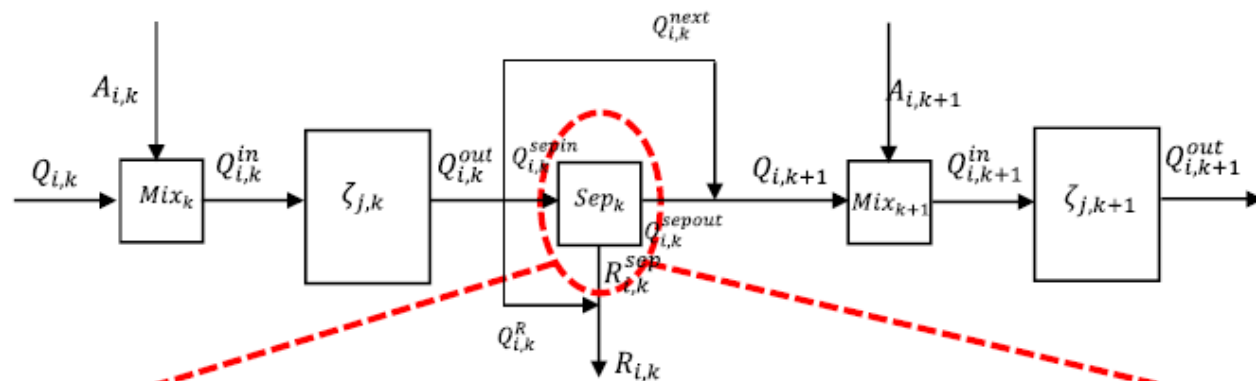


Superstructure-based Methodology





Superstructure-based Methodology





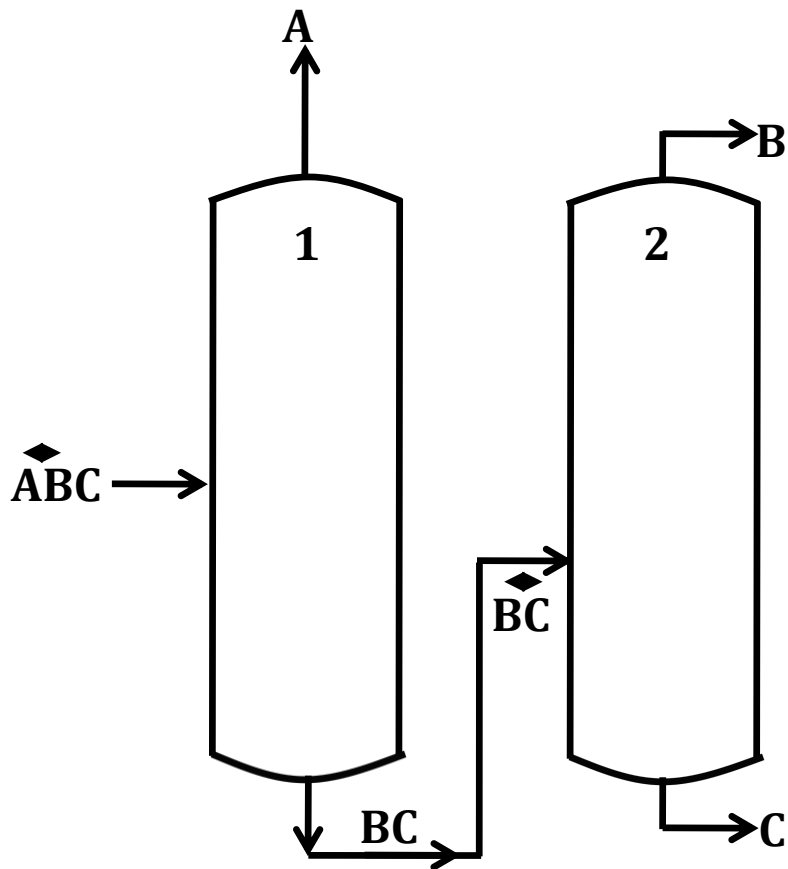
- Operating conditions (temperature, pressure, catalyst)
- One reaction can occur (not one of the parallel reactions; binary)
- Mass balance based on conversion data w.r.t. a reference component (binary)
- Parallel reactions must occur simultaneously (binaries)
- Mass balance based on conversion data and selectivity between reactions (binaries)
- One stream can leave the splitter



Distillation Column Basics



Example: Distillation of a mixture of 3 components A, B & C

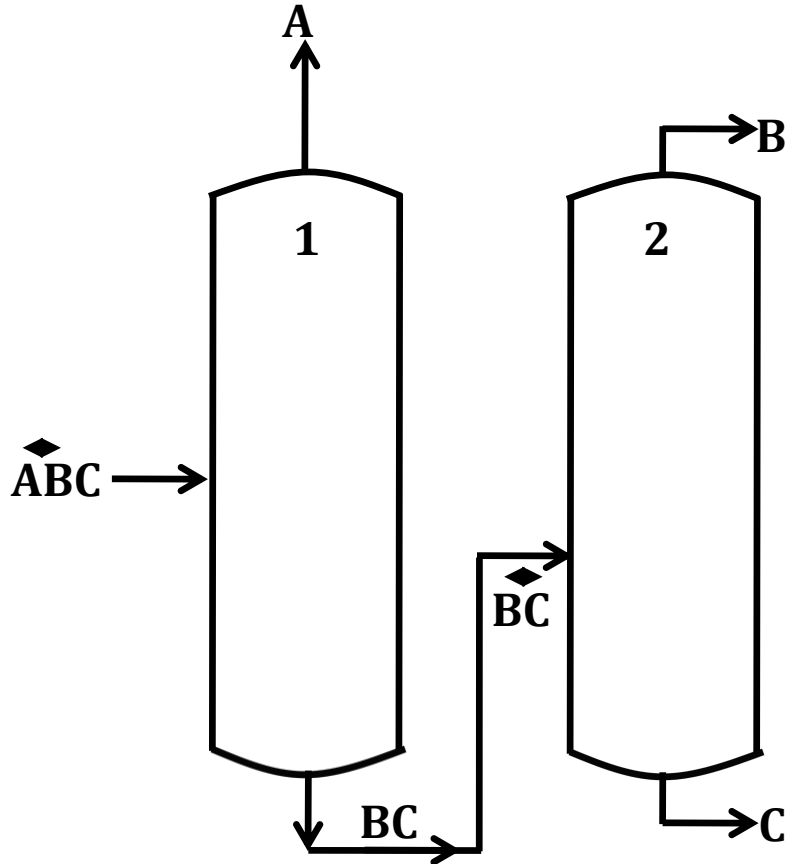




Distillation Column Basics



Example: Distillation of a mixture of 3 components A, B & C



A: Light key

B: Light key

B: Heavy key

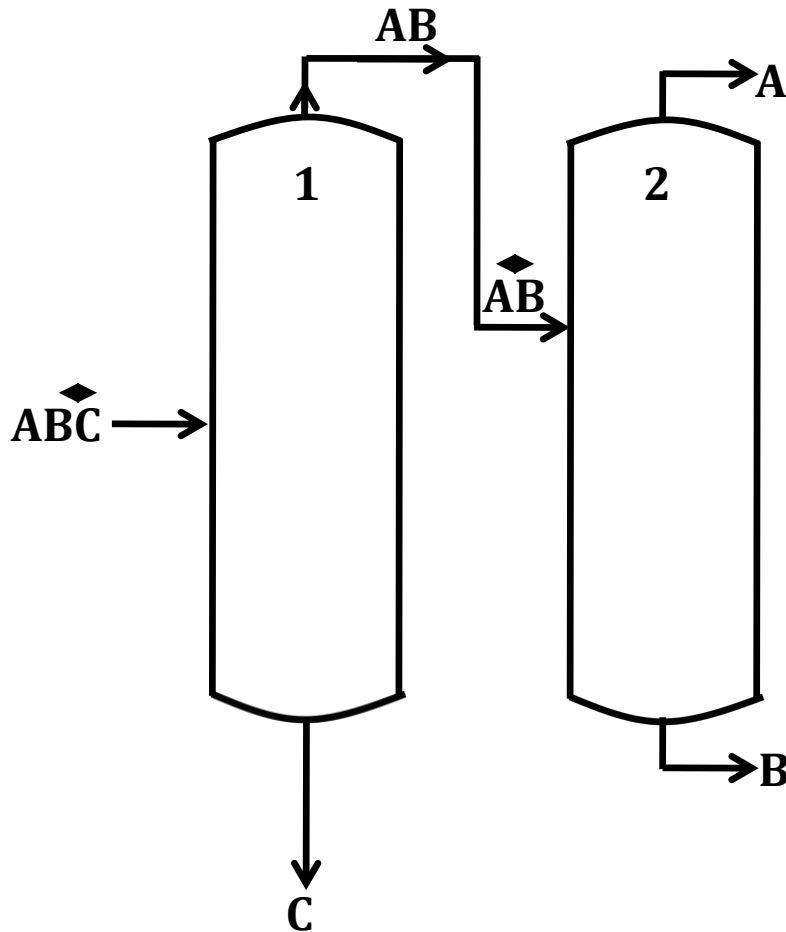
C: Heavy key



Distillation Column Basics



Example: Distillation of a mixture of 3 components A, B & C

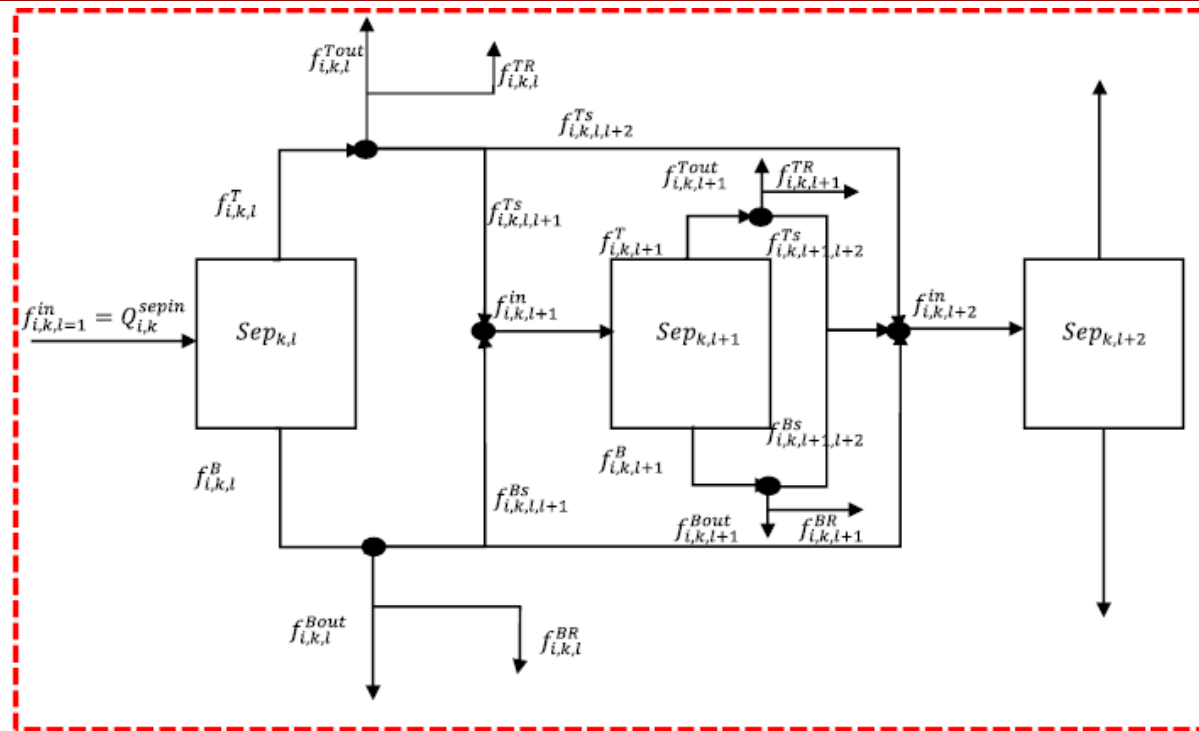


B: Light key
C: Heavy key

A: Light key
B: Heavy key



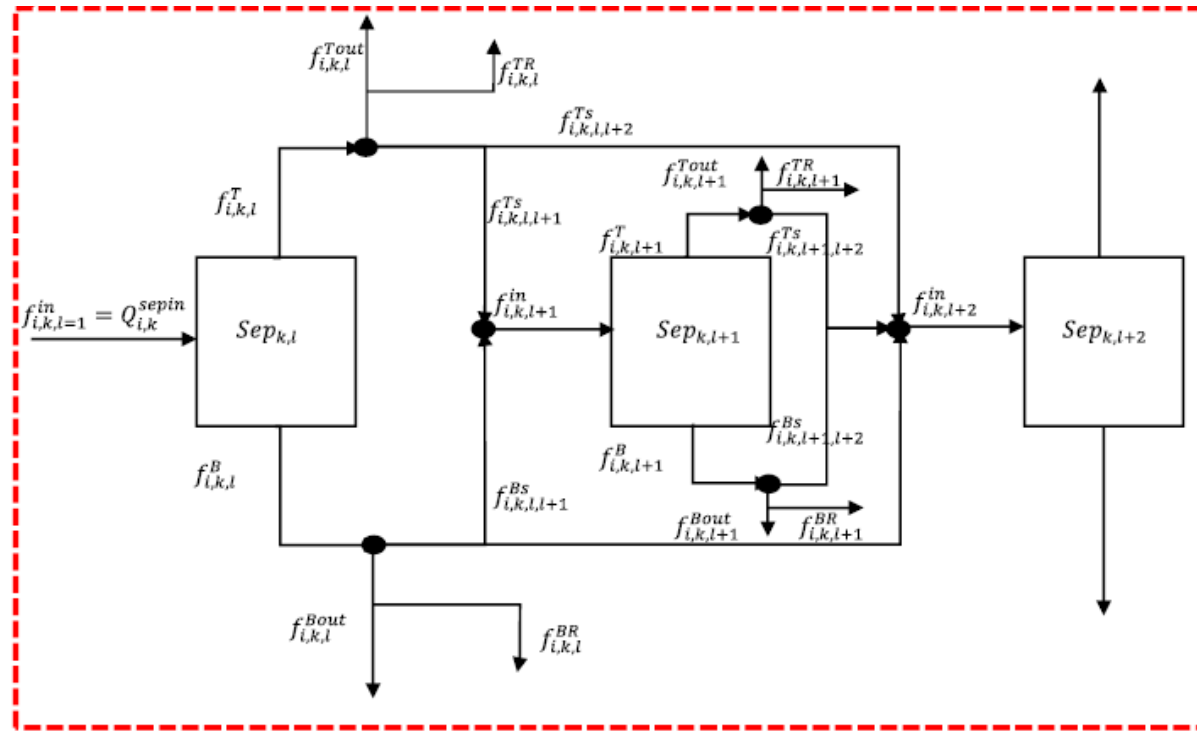
Distillation Columns Modeling



- Each column: 1 light key, 1 heavy key
- Any component: light/heavy key in only one column
- Detect the present inlet components
- Light/heavy key from among present inlet components
- Components lighter than light key -> top product



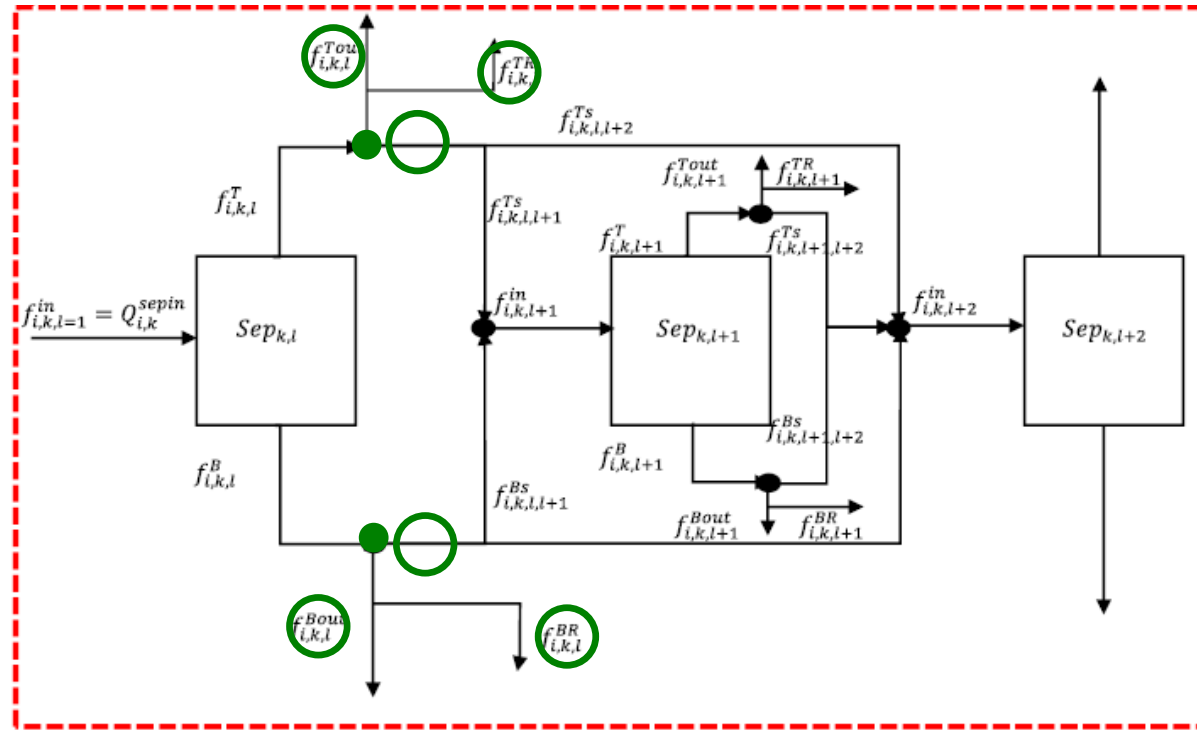
Distillation Columns Modeling



- Components heavier than heavy key -> bottom product
- Heavy key: component with the next highest BP than that of the light key
- A prior column absent => all subsequent columns are absent
- Mass balance



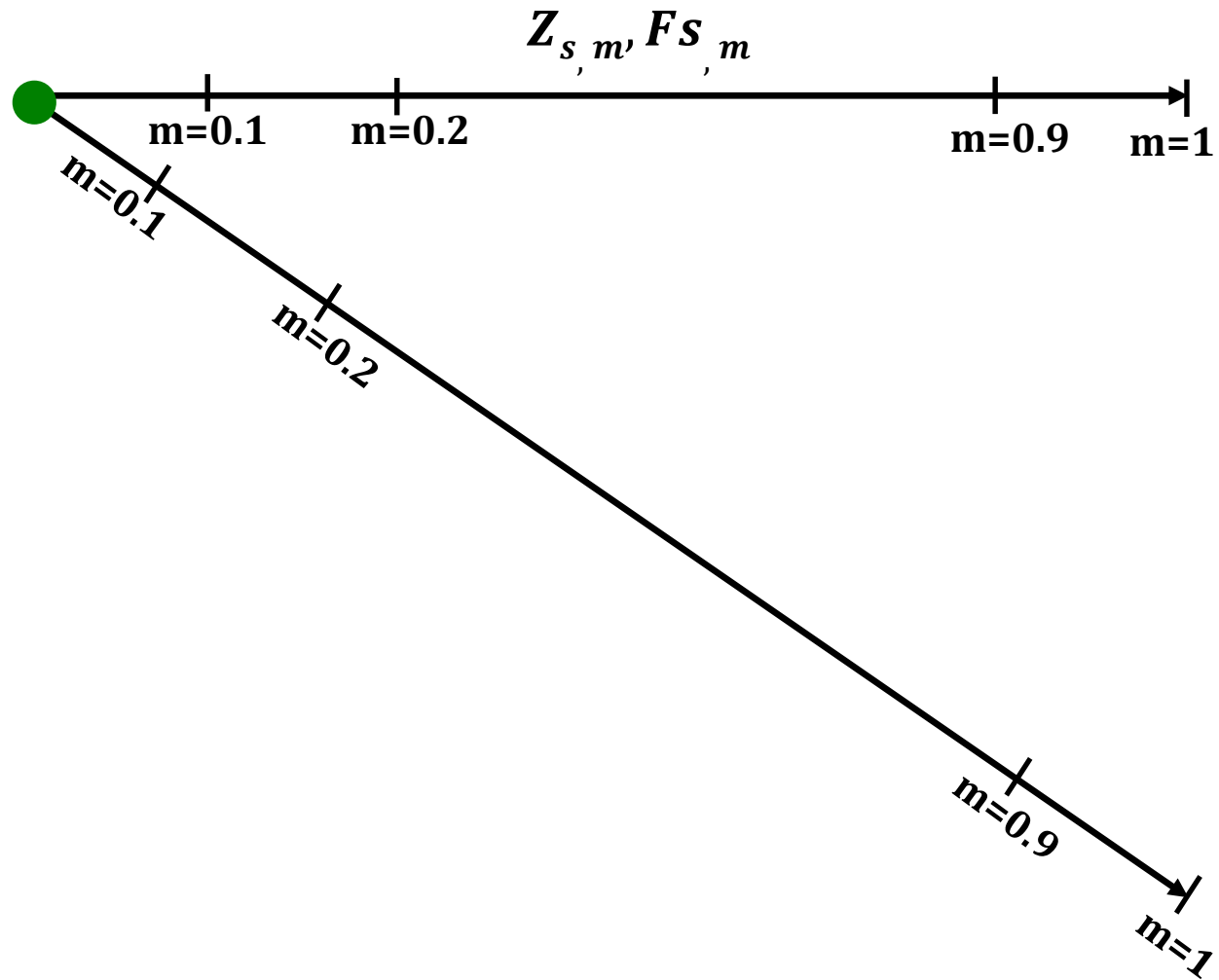
Splitter Modeling



-Splitters associated with distillate/bottoms: non linear constraints

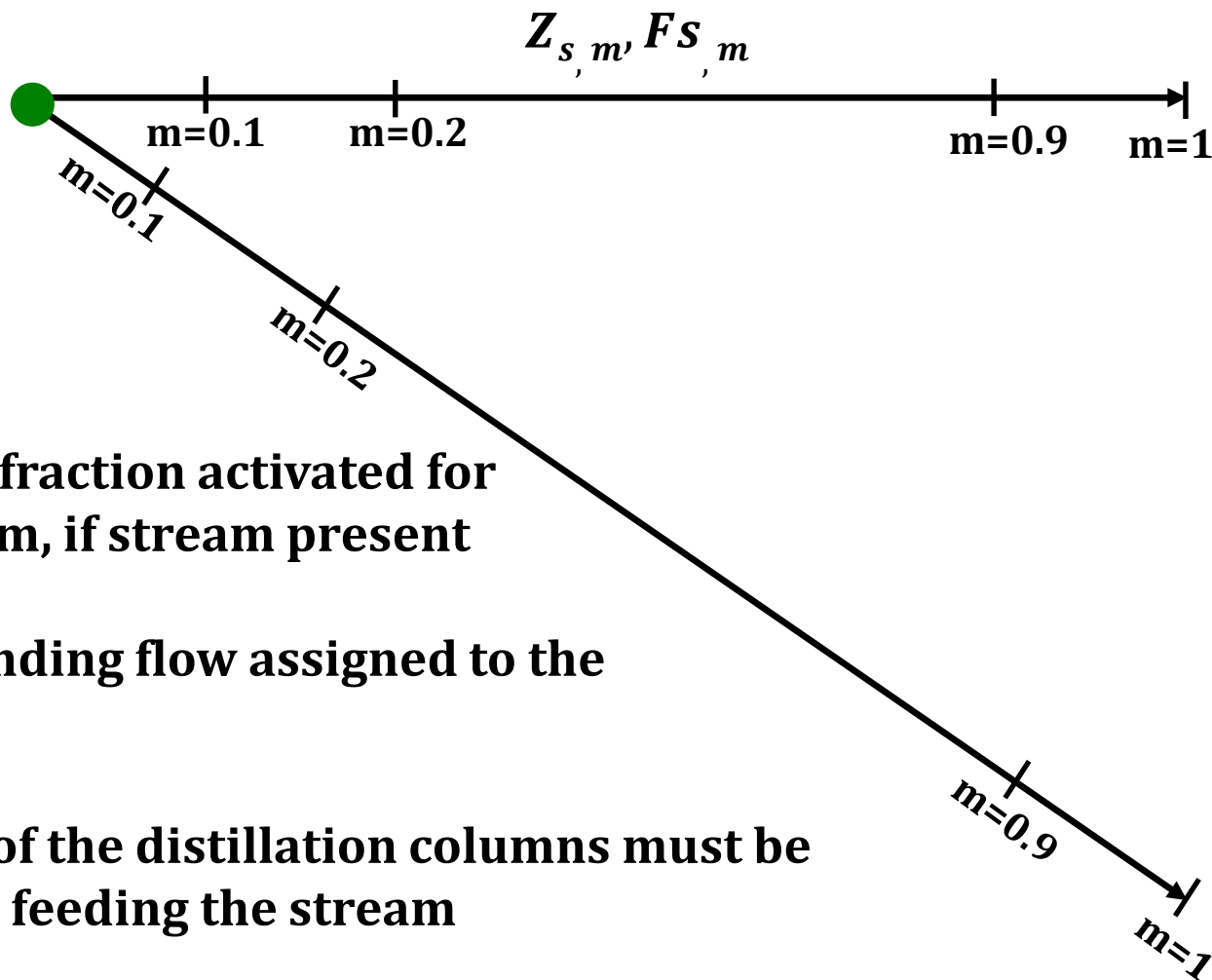


Splitter Modeling





Splitter Modeling



-One split-fraction activated for each stream, if stream present

-Corresponding flow assigned to the Stream

-Only one of the distillation columns must be chosen for feeding the stream

-Sum of the chosen split-fractions=1



Economic Considerations



-Revenue from products

-Raw material costs

-Annualized separation cost (column) = $\frac{\beta(\text{Total feed flowrate to the column})}{\text{BP difference b/w the light and heavy key}}$

-Profit

-MILP formulation



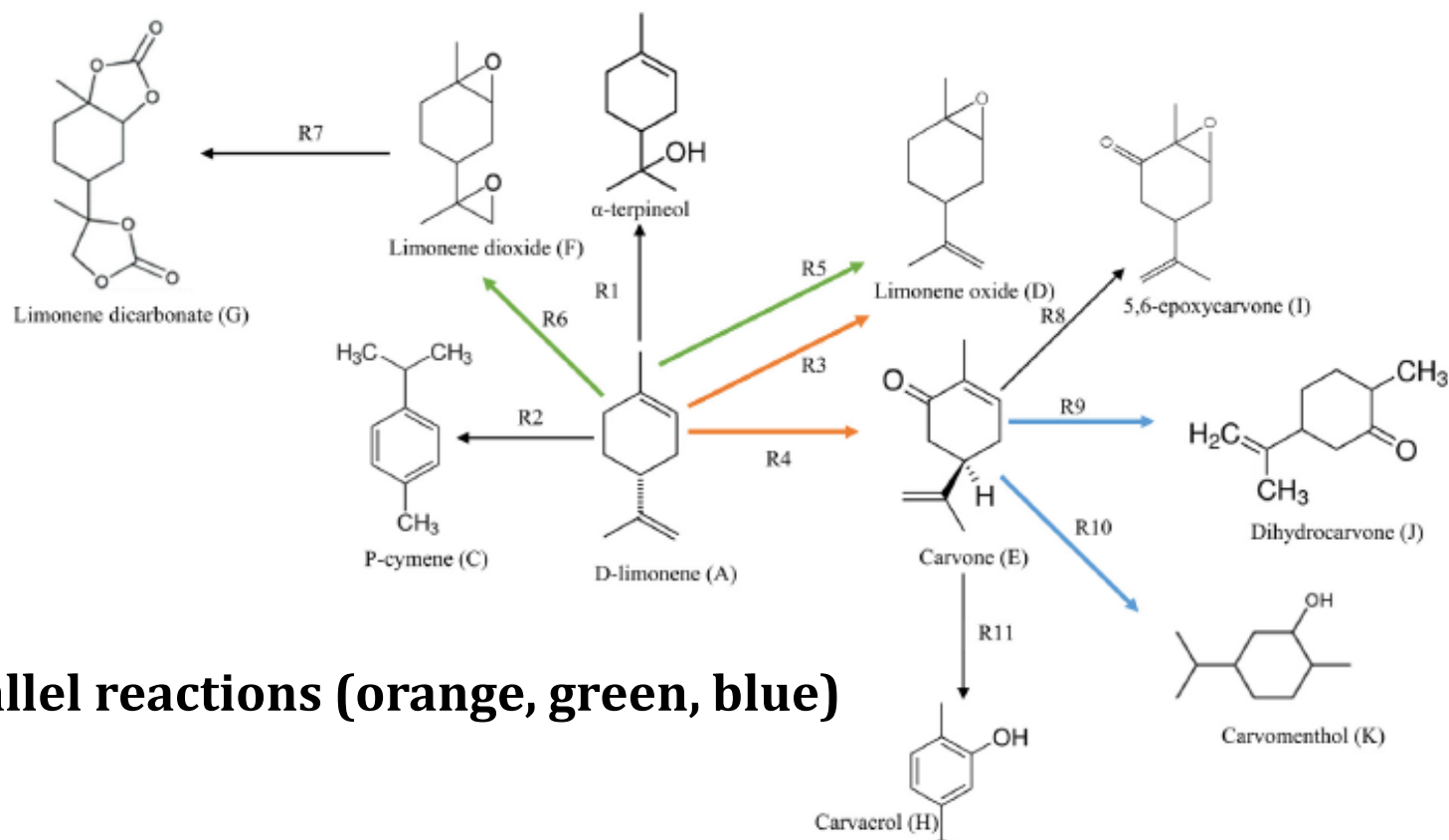
- D:limonene as starting material**
- Worldwide production of d-limonene \geq 70,000 t in 2013**
- Assumption: plant processing 15,000 t limonene per annum**



-D:limonene as starting material

-Worldwide production of d-limonene $\geq 70,000$ t in 2013

-Assumption: plant processing 15,000 t limonene per annum



-Parallel reactions (orange, green, blue)



Case Study: Reaction Network Data



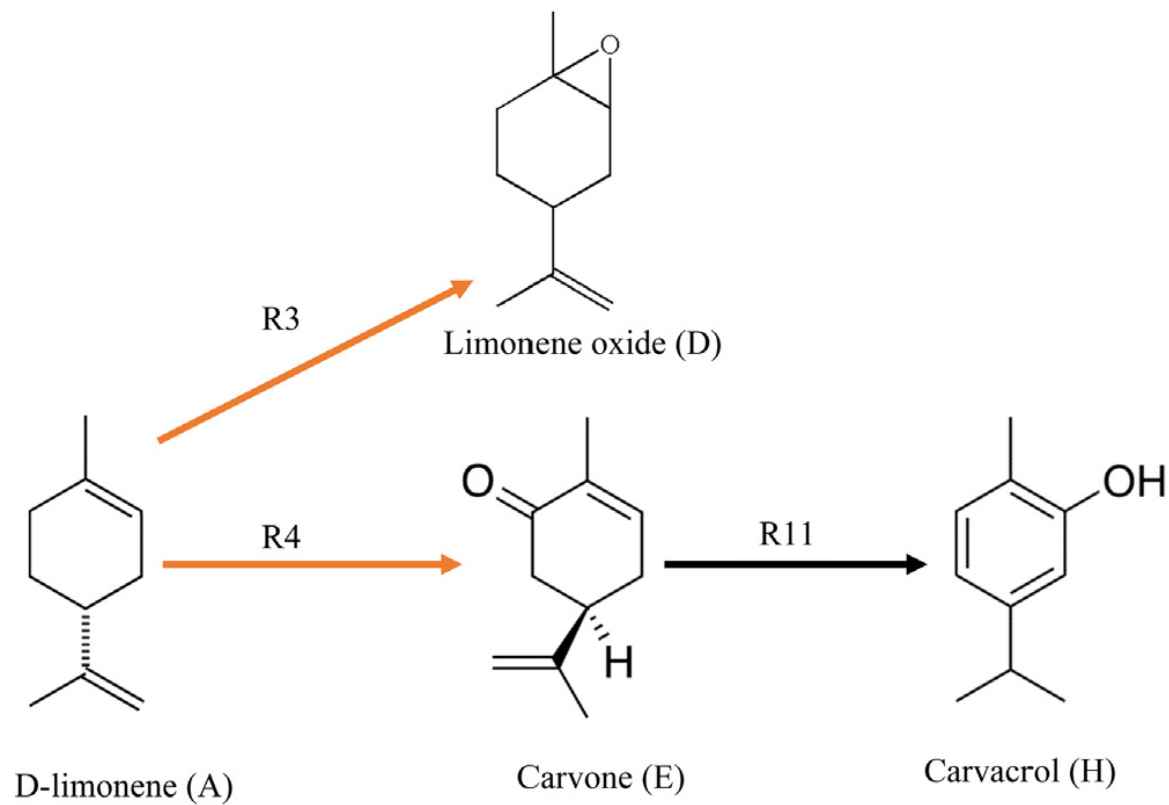
| Reaction | Conversion | Selectivity (mol%) | Conditions | Source |
|----------|------------|--------------------|---------------|------------|
| R1 | 93 | – | 25 °C, 1atm | Literature |
| R2 | 100 | – | 170 °C, 1atm | Literature |
| R3 | 100 | 59.5 | 75 °C, 1atm | Literature |
| R4 | 100 | 40.5 | 75 °C, 1atm | Assumed |
| R5 | 98 | 74.5 | 40 °C, 1atm | Literature |
| R6 | 98 | 25.5 | 40 °C, 1atm | Assumed |
| R7 | 100 | – | 140 °C, 30bar | Literature |
| R8 | 32 | – | 50 °C, 1atm | Literature |
| R9 | 90 | 62.6 | 100 °C | Literature |
| R10 | 90 | 37.4 | 100 °C | Assumed |
| R11 | 80 | – | – | Literature |

-Boiling points: from databases

-Market price of certain molecules: group contribution method

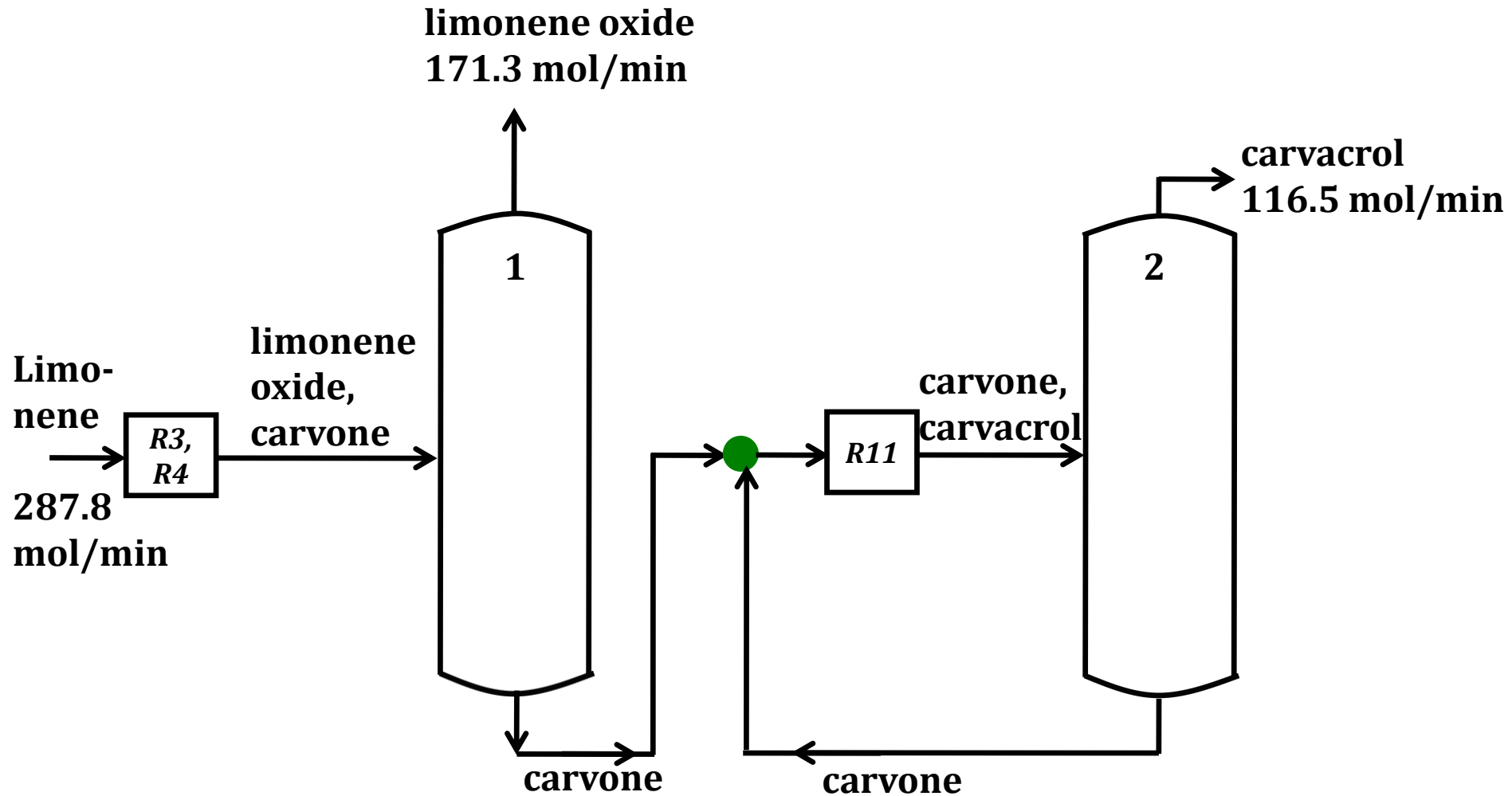


Optimal Reaction Network



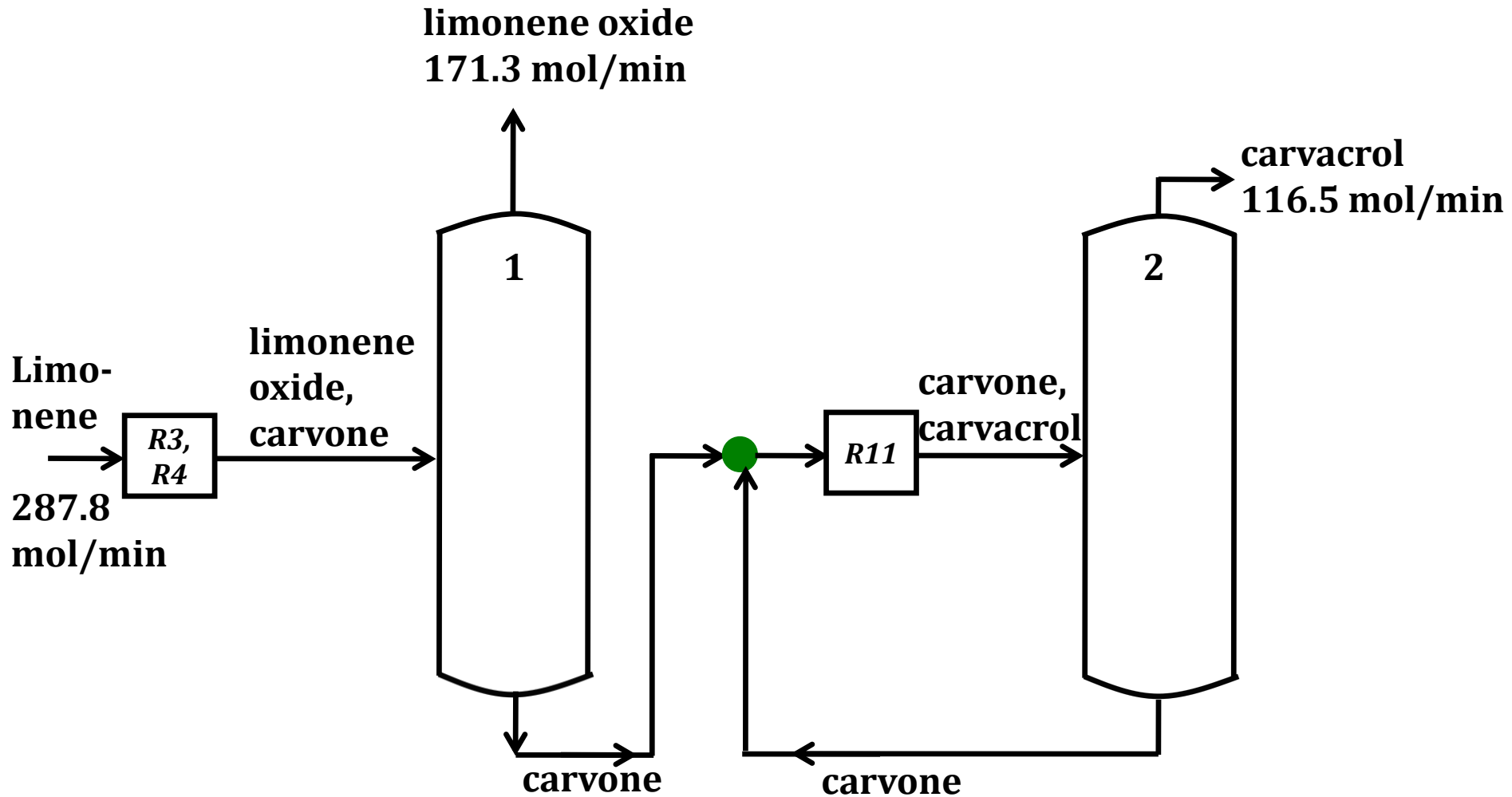


Optimal Process Flowsheet





Optimal Process Flowsheet



Profit: \$20.8 million per annum



Conclusions



- Process synthesis from reaction networks: a superstructure-based approach**
- Reactor model**
- Distillation sequence model**
- Splitter model**
- Demonstrated the framework on a case study**



Thank You!